

Capillary Gas Chromatography-Mass Spectrometry Determination of Pesticide Residues in Vegetables

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Abstract: In this study, a gas chromatography-mass spectrometry method is successfully developed for the determination of pesticide residues monocrotophos, chlorpyrifos and endosulfan in cauliflower and capsicum. The samples were extracted with ethyl acetate, cleaned up and purified through solid-phase extraction with Florisil and activated charcoal. Experiments on two fortification concentrations are carried out and the limits of detection are 0.005, 0.007 and 0.002 mg kg⁻¹ for monocrotophos, chlorpyrifos and endosulfan respectively. The average recoveries of pesticide residues in cauliflower and capsicum samples are 89.0 to 110.0 %.

Key words: *Monocrotophos · Chlorpyrifos · Endosulfan*

INTRODUCTION

India produces about 109 million metric tons of vegetables and it is the second largest producer after China and accounts for 13.4% of world production. Surveys carried out by institutions spread throughout the country indicate that 50-70% of vegetables are contaminated with insecticide residue [1]. The use of pesticides to control pest and diseases are a common practice in the fields to increase crop yield. However, these chemicals can reach plant tissues, leaving residues that can be detected in the vegetables. This may become a significant route to human exposure to these toxic compounds. In order to protect consumer's health, maximum residue levels (MRLs) in these vegetables have been established in different countries and internationally by Codex Alimentarius. The high number of pesticides to be monitored in those matrices, along with the typically low concentrations of the MRLs, requires highly sensitive and selective methods. Consequently, sample preparation becomes a key step of the analytical procedure. In recent times, extensive efforts have been made to the development of new sample preparation techniques that save time, labor and solvent consumption to improve the analytical performance of the procedure. Analytical instruments are needed to determine, quantify and confirm pesticide residues in vegetables for both research and regulatory purposes. The pesticides are generally

analysed by spectrophotometry [2-4], thin layer chromatography (TLC) [5-7], high performance liquid chromatography (HPLC) [8-10], gas chromatography (GC) [11-14] and GC-MS [15-20]. The present study describes method of extraction, cleanup and determination of pesticides by using gas chromatography (GC) equipped with mass detector (MS) for the separation, identification and quantification of monocrotophos, chlorpyrifos and endosulfan on cauliflower and capsicum were developed and validated. Finally, the method was applied to the determination of these pesticides in commercial samples collected from the local markets. Therefore, the purpose of this study was to develop an analysis scheme for determination of these pesticides in cauliflower and capsicum GC-MS.

Experimental

Chemical and Reagents: The organic solvent ethyl acetate and hexane used were HPLC grade and purchased from E Merck. Technical grade pesticide standards were used for standardisations. The standards were stored in a freezer at -5°C. Anhydrous sodium sulphate (AR) from E Merck used for residue extraction was maintained at 300°C overnight and kept in air tight container.

Sample Preparation: The fresh cauliflower and capsicum samples were taken for the extraction of pesticide residues. Each vegetable was chopped into small

pieces; a representative sample (50gm) was macerated with 5-10gm anhydrous sodium sulphate in blending machine to make fine paste. The macerated sample was extracted with 100 ml of ethyl acetate on mechanical shaker for 1 h; extract was filtered and clean up.

Sample Clean up: The clean-up of monocrotophos, chlorpyrifos and endosulfan was carried out by using column chromatography. Column (60cm x 22mm) was packed with, Florisil and activated charcoal (5:1 w/w) in between the two layers of anhydrous sodium sulphate. Extract was eluted with 125 ml mixture of ethyl acetate and hexane (3:7 v/v). Elute was concentrated to 5 ml on rotary evaporator and injected into GC-MS.

Standard Preparation: For preparation of stock solution, standards were dissolved in ethyl acetate and four levels of intermediate standard solution of each pesticide were prepared maintaining the same matrix concentration for the preparation of calibration curve and stored at -4°C in the dark. Working solutions were prepared daily by appropriate dilution with ethyl acetate.

Instrumentation: GC-MS analysis was performed with a Varian 3800 gas chromatograph with electronic flow control (EFC) and fitted with a Saturn 2200 ion-trap mass spectrometer (Varian Instruments, Sunnyvale, CA, USA). Samples were injected into a Varian 8200 auto sampler SPI / 1079 split / splitless programmed-temperature injector using a 10 μ l syringe operated in the large volume injection technique. The glass liner was equipped with a plug of carbofrit (Resteck, Bellefonte, PA, USA). A fused-silica untreated capillary column 2 m 0.25 mm I.D. from Supelco (Bellefonte, PA, USA) was used as a guard column connected to a Rapid-MS [wall-coated open tubular (WCOT) fused-silica CP-Sil 8 CB low bleed of 10m 0.53 mm I.D., 0.25 mm film thickness] analytical column from Varian Instruments (Sunnyvale, CA, USA) for high speed analysis. The mass spectrometer was operated in electron impact (EI) ionization mode. The computer that controlled the system also held a GC-MS library specially created for the target analytes under our experimental conditions. The mass spectrometer was calibrated weekly with perfluoro-tributylamine. Helium (99.999%) at a flow-rate of 1 ml min⁻¹ was used as carrier and collision gas.

Instrumental Conditions: Sample aliquots of 10 ml were injected into the GC operating at a syringe injection flow-

rate of 10 ml s⁻¹. The initial injector temperature of 70°C was held for 0.5 min and then increased at 100°C min⁻¹ to 310°C, which was held for 10 min. After injection the column temperature, initially 70°C, was held for 3.5 min, then increased at 50°C min⁻¹ to 150°C, then increased at 3°C min⁻¹ to 235°C and finally raised to 300°C at 50°C min⁻¹ and held for 3 min. The ion-trap mass spectrometer was operated in EI-MS mode. The transfer line, manifold and trap temperatures were 280, 50 and 200°C, respectively. The analysis was performed with a filament-multiplier delay of 4.75 min to prevent instrument damage. The automatic gain control (AGC) was activated with an AGC-target of 5000 counts. The emission current for the ionisation filament was set at 80 mA, generating electrons with an energy of 70 eV. The axial modulation amplitude voltage was 4.0 V.

RESULTS AND DISCUSSION

Validation of the Method: In order to check the feasibility of the GC-MS method for the analysis of pesticide residues in fresh vegetable sample extracts, it was validated using cauliflower and capsicum extracts.

Identification and Confirmation of Target Analytes: The identification of the pesticides was based on the retention time windows (RTW) that are defined as the retention time average \pm 3 S.Ds of the retention time when 10 blank samples spiked at the second calibration level of each compound were analysed. The confirmation of a previously identified compound was done by comparing the GC-MS spectra obtained in the sample with another stored as reference spectrum in the same experimental conditions. The reference spectra were obtained daily by injecting a blank cauliflower and capsicum sample spiked at the concentration of the second calibration point.

Identification and Quantification: The compound was identified by comparing its retention time with respect to technical grade reference standard. The quantitative determination was carried out with the help of a calibration curve drawn from chromatographic experiments with standard solution. For quantification an external calibration curve with four different concentrations of each pesticide, with matrix matching were made. The standard solutions for the calibration curves were prepared in control matrix because samples may possess co-extractants in the matrix which may affect the peak area of the unknown samples.

Table 1: Molecular formula, retention time, LODs and LOQs of monocrotophos, chlorpyriphos and endosulfan

Compound	Molecular formula	RT (min)	LoDs (mg kg ⁻¹)	LoQs (mg kg ⁻¹)
Monocrotophos	C ₇ H ₁₄ NO ₃ P	17.89	0.005	0.015
Chlorpyriphos	C ₉ H ₁₁ Cl ₃ NO ₃ PS	25.12	0.007	0.021
Endosulfan	C ₉ H ₆ Cl ₆ O ₃ S	26.72	0.002	0.006

Table 2: Quantitation ion, conformation ion and calibration range of monocrotophos, chlorpyriphos and endosulfan

Compound	Quantitation ion	Confirmation ion	Calibration range (mg kg ⁻¹)	Correlation coefficient	Coefficient of variation (n = 5) %
Monocrotophos	127	98	0.02-1.00	0.991	5.8
Chlorpyriphos	97	314	0.02-1.00	0.988	6.0
Endosulfan	373	237	0.02-1.00	0.989	5.8

Table 3: Recovery of pesticides in the spiked samples.

Sample	Compound	Concentration (mg kg ⁻¹)	Recovery (%)	Coefficient of variation (n = 5) %
Cauliflower	Monocrotophos	1.0	89.80	4.58
Cauliflower	Chlorpyrifos	1.0	99.80	3.90
Cauliflower	Endosulfan	1.0	108.20	4.40
Capsicum	Monocrotophos	1.0	91.20	4.25
Capsicum	Chlorpyrifos	1.0	100.30	4.56
Capsicum	Endosulfan	1.0	110.00	4.68

Table 4: Amounts of pesticides residue detected in cauliflower and capsicum samples.

Sample	Monocrotophos (mg kg ⁻¹)	Chlorpyriphos (mg kg ⁻¹)	Endosulfan (mg kg ⁻¹)
Cauliflower	nd	nd	0.002
Cauliflower	0.024	nd	0.003
Cauliflower	nd	0.002	0.001
Cauliflower	0.027	nd	nd
Cauliflower	nd	nd	0.002
Capsicum	nd	0.012	0.002
Capsicum	0.021	nd	0.021
Capsicum	0.024	0.008	0.027
Capsicum	0.018	0.003	nd
Capsicum	0.020	nd	0.026

nd = non detected

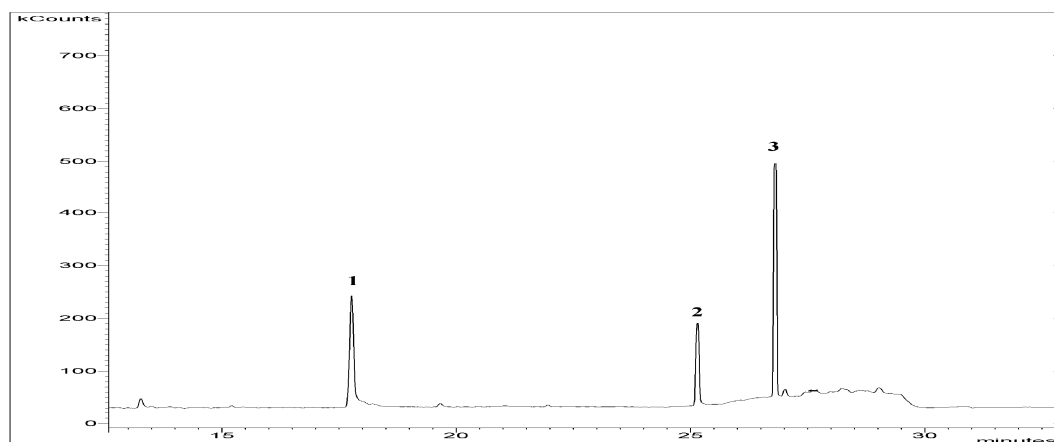


Fig. 1: TIC of monocrotophos¹, chlorpyriphos² and endosulfan³ in actual sample of capsicum.

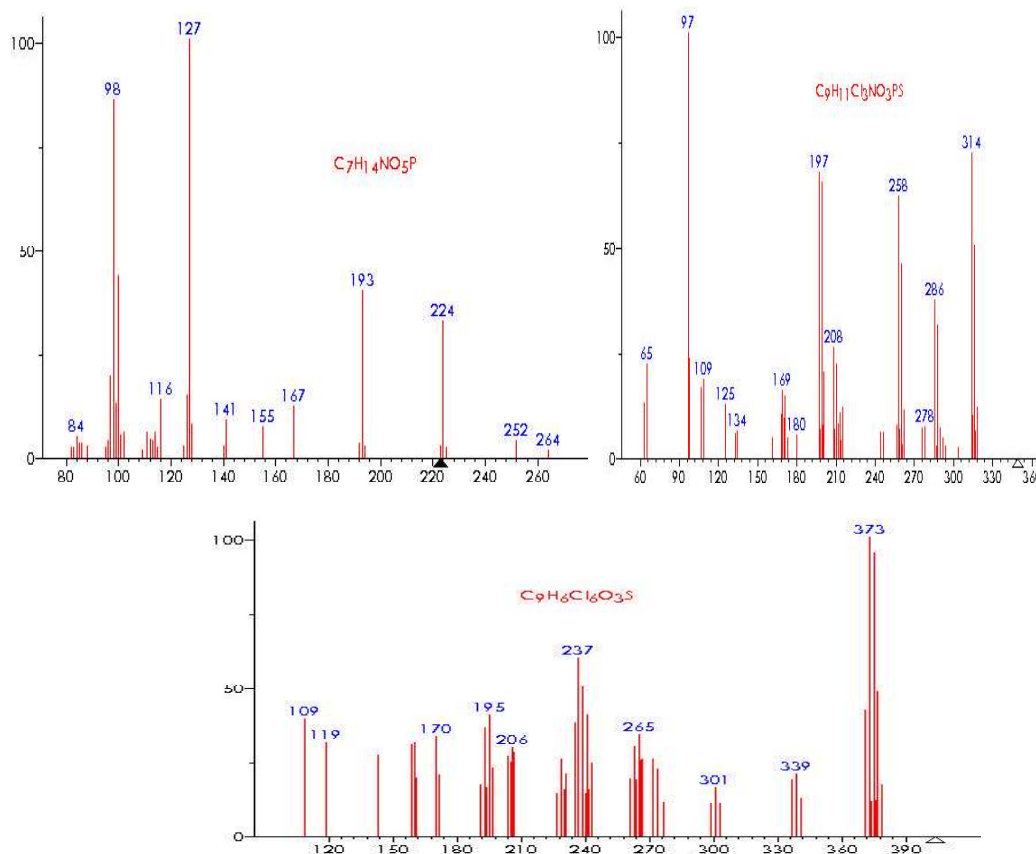


Fig. 2: Mass spectra of monocrotophos, chlorpyrifos and endosulfan.

Limit of Detection and Limit of Quantification: The limit of detection (LoD) was calculated from the peak intensity at 0.01 mg kg^{-1} and blank in recovery tests. LoD was defined as $S/N > 4$ so that it is in the linear range of the standard calibration. The LoD of monocrotophos, chlorpyrifos and endosulfan was 0.005 , 0.007 and 0.002 mg kg^{-1} respectively. LoQ was obtained for monocrotophos, chlorpyrifos and endosulfan was 0.015 , 0.021 and 0.006 mg kg^{-1} respectively (Table 1). Linear calibration curves were found between peak areas and analyte concentration in the whole range of studies. The linear regression ($y = a + bx$) parameters for method calibration were taken (Table 2). The correlation coefficient of analytical curves were near 0.99 , with linearity for each compound, which allows the quantitation of these compounds by the method external standardization.

Recovery: Recovery studies were performed to examine the efficacy of extraction and clean up. Untreated cauliflower and capsicum samples were spiked with known concentration of the pure pesticides standard

solution and extraction and clean-up were performed as described earlier. The concentration of each pesticide in the final extracts was calculated (Table 3). The average recoveries of pesticide residues in cauliflower and capsicum samples are 89.0 to 110.0% .

Application to the Analysis of Market Samples: In order to test the feasibility of the GC-MS approach for routine analysis of pesticide residues in the market samples of vegetables (cauliflower and capsicum) were analysed for the target compounds. The concentrations of each pesticide in the final extracts of the market were obtained and calculated (Table 4). Figure 1 shows TIC of monocrotophos, chlorpyrifos and endosulfan in actual sample of capsicum and Figure 2 shows mass spectra of monocrotophos, chlorpyrifos and endosulfan.

CONCLUSION

In this study, the operating parameters of GC-MS for the analysis of 3 representative pesticides in cauliflower and capsicum were optimized and sample preparation

method was evaluated. The main conclusions of the study can be summarised as follows: (i) a good separation and high sensitivity was achieved by GC-MS method for all pesticides using a capillary column, (ii) the classical procedure that involves extraction with ethyl acetate, partitioned from the aqueous matrix using anhydrous sodium sulphate a cleanup with florisil and activated charcoal, showed an efficient removal of interferences, providing a simple, rapid and reliable analysis of pesticides in all matrices; (iii) for most of the pesticides assayed the performance characteristics obtained within validation study were acceptable, within the quality control requirements. Applying this method, analysis time is shorter compared to other methods. Thus, high sample throughput can, therefore, be achieved, which is useful in pesticide monitoring programs with a large number of samples to analyse.

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