

Identification of Nonpolar Chemical Composition *Spartium junceum* flower Growing in Iran by GC-MS

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Abstract: The *Spartium junceum* is a medicinal plant that belongs to the fabacea family. The nonpolar chemical composition of it was investigated and fifty nine compounds were identified by GC-MS. The main constituents were the n-Hexadecanoic acid (14.27%), 9, 12, 15-Octadecatrien-1-ol(13.07%), Tetradecanoic acid (6.59%) Octadecanoic acid (3.68%) and γ -Sitosterol(3.67%).

Key words: Chemical composition, N-Hexane extraction, N-Hexadecanoic Acid, GC-MS, *Spartium junceum*

INTRODUCTION

Medicinal plants are believed to be an important source of new chemical substances with potential therapeutic effects. Thus study of plant species that traditionally have been used as pain killer should still be seen as a strategy in research for new analgesic drugs. The *Spartium junceum* belongs to the Fabaceae family, is a small shrub indigenous in the Mediterranean countries and cultivated as an ornamental plant. Its flowers are large, yellow and by greenish scent [1]. *S. junceum* contains flavonoids and saponins. Two new flavonoids and saponin were isolated from its flowers in a study by Bilia *et al.* [2]. Another study showed that *S. junceum* seeds contain a large amount of lectin [3]. An herbal tea (known as Zahraa) widely consumed in Syria contains 6-14 species components including *S. junceum* [4].

Flowers of *S. junceum* are used for the treatment of gastric ulcers in Turkish folk medicine [5]. It has been shown in a study by Yesilada *et al.*, [6] that *S. junceum* did not have anti-helicobacter pylori activity. Flavonoids-rich fractions from the flowers *S. junceum* showed potent antioxidant activity reported by Yesilada *et al.* [7].

It is known to simulate uterine contractions and GI tract, help body to dispose excess fluid by increasing amount of urine and cause vomiting [8]. It has been studied for its antifertility activity in mammalian male [9].

Another study showed injection of *S. junceum* extract in adult male rats reduce the rate of fertility and acrosin enzyme activity [10]. Only one study reported the antinociceptive and anti-inflammatory effects of this plant [11]. Keeping this in view, the present study has been done to identify nonpolar compounds of n-hexane extract of this plant.

MATERIALS AND METHODS

The experimental protocol used in this study was approved by the research committee of North Khorasan Payame Noor University, Bojnourd, Iran.

Plant Material: Flower of *Spartium junceum* were collected at the flowering stage from the Bojnourd, Iran, in Jun 2011 and identified at the Research Center for Plant Sciences, Ferdowsi University of Mashhad, Iran. A voucher specimen has been deposited in the Environmental Department of Bojnourd Herbarium (EDBH:00107).

Isolation of Chemical Compounds: Flowers of *Spartium junceum* were air-dried for 3 days before isolation of chemical composition. The plant material (100g) was cut into small pieces. The plant powder was macerated in 95% pure n-hexane as a solvent for 48 hours, filtered through a Whatman paper, then evaporated off the solvent in vacuum by rotary evaporator to yield light

yellow oil and dried over by adding anhydrous Na_2SO_4 . In absolute oil recovery, concentrate oil was dissolved in minimum volume of absolute alcohol to remove the natural waxes present in the essential oil. It was kept at -14°C for 48 hrs and then it was filtered through a filter paper. Alcohol was removed by distillation and traces of alcohol were removed by passing nitrogen gas through it [12].

Gas Chromatography and Mass Spectrometry:

Gas chromatographic analysis was performed on an Hewlett-Packard(HP)6890A instrument equipped with a flame ionization detector and Rtx-5MS ($15\text{ m} \times 0.25\text{ mm} \times 0.25\text{ }\mu\text{m}$) capillary column, while the essential oil components were identified on an Agilent Technologies 5973N mass spectrometer. The GC settings were as follows: the initial oven temperature was held at 35°C for 6 min and ramped at 5°C min^{-1} to 150°C for 0 min and then ramped at $10^\circ\text{C min}^{-1}$ to 280°C for 3 min. The injector temperature was maintained at 250°C . The samples ($1\text{ }\mu\text{L}$) were injected, with a split ratio of 1:10. The carrier gas was helium at flow rate of 1.0 mL min^{-1} . Spectra were scanned from 20 to 550 m/z at 2 scans s^{-1} . Most constituents were identified by gas chromatography by comparison of their retention indices

with those of the literature or with those of authentic compounds available in our laboratories. The retention indices were determined in relation to a homologous series of *n*-alkanes under the same operating conditions. Further identification was made by comparison of their mass spectra on both columns with those stored in NIST 05 and Wiley 275 libraries or with mass spectra from literature [13-18]. Component relative percentages were calculated based on GC peak areas without using correction factors.

RESULTS AND DISCUSSION

The average yield of chemical composition of the leaves of *Spartium junceum* was about 0.58%. Table 1 reports the chemical composition of the phytochemical components under study. Fifty nine components were identified, accounting for %76.27 of the total oil.

The various compounds were identified by comparison of their Kováts retention indexes, determined utilizing a non-logarithmic scale on non-polar (Rtx-5MS) columns and by comparison of the mass spectra of each GC component with those of standards and with reported data [19].

Table 1: Percentage composition of the essential oil isolated from aerial parts of *spartium junceum*

NO	compound	Experimentally determined K_i^a	HP GC-MS Peak area [%]	Method of identification
1	E thyl benzene	850	0.28	GC-MS,Ms
2	1,3-dimethyl-benzene	866	0.29	GC-MS,Ms
3	Mesitylene	995	0.14	GC-MS,Ms
4	Decane	1000	0.13	GC-MS,Ms
5	Phenylethyl Alcohol	1107	0.25	GC-MS,Ms
6	Benzothiazole	1216	0.17	GC-MS,Ms
7	2-Propenoic acid, 3-phenyl-, methyl ester	1380	0.11	GC-MS,Ms
8	Pentadecane	1499	0.14	GC-MS,Ms
9	2,3,6-trimethyl-Naphthalene	1550	0.11	GC-MS,Ms
10	Dodecanoic acid	1564	1.64	GC-MS,Ms
11	Dodecanoic acid, ethyl ester	1592	0.47	GC-MS,Ms
12	Hexadecane	1598	0.44	GC-MS,Ms
13	Pentadecane, 2,6,10,14-tetramethyl-	1649	0.21	GC-MS,Ms
14	Heptadecane	1663	0.08	GC-MS,Ms
15	2,6,10,14-tetramethyl-Pentadecane	1705	0.35	GC-MS,Ms
16	Phenol, 2-(1-phenylethyl)-	1721	0.41	GC-MS,Ms
17	Tetradecanoic acid	1763	6.59	GC-MS,Ms
18	Phenanthrene	1784	0.4	GC-MS,Ms
19	Tetradecanoic acid, ethyl ester	1792	1.47	GC-MS,Ms
20	Octadecane	1798	0.75	GC-MS,Ms
21	Hexadecane, 2,6,10,14-tetramethyl-	1808	0.65	GC-MS,Ms
22	Tetradecanoic acid,trimethylsilyl ester	1850	0.16	GC-MS,Ms
23	Pentadecanoic acid	1857	0.21	GC-MS,Ms
24	Dibenzothiophene, 3-methyl-	1862	0.19	GC-MS,Ms
25	1,2-Benzenedicarboxylic acid, bis(2-methylpropyl) ester	1869	0.45	GC-MS,Ms
26	Nonadecane	1897.4	0.54	GC-MS,Ms
27	1,2-Benzenedicarboxylic acid, butyl 2-ethylhexyl ester	1915	0.11	GC-MS,Ms
28	Hexadecanoic acid, methyl ester	1924	0.3	GC-MS,Ms
29	n-Hexadecanoic acid	1964.15	14.27	GC-MS,Ms

Table I: Continue

NO	compound	Experimentally determined Ki ^a	HP GC-MS Peak area [%]	Method of identification
30	Hexadecanoic acid, ethyl ester	1992.4	2.34	GC-MS,Ms
31	Eicosane	1999	0.48	GC-MS,Ms
32	Cyclic octaatomic sulfur	2055	0.61	GC-MS,Ms
33	1-Octadecanol	2082	0.18	GC-MS,Ms
34	14-beta -H-PREGNA	2091	0.14	GC-MS,Ms
35	Heneicosane	2097	0.51	GC-MS,Ms
36	E-15-Heptadecenal	2104	0.08	GC-MS,Ms
37	Methyl stearate	2024.7	0.05	GC-MS,Ms
38	9,12,15-Octadecatrien-1-ol	2042	13.07	GC-MS,Ms
39	Octadecanoic acid	2161	3.68	GC-MS,Ms
40	9,12,15-Octadecatrienoic acid, ethyl ester, (Z,Z,Z)-	2169	2.05	GC-MS,Ms
41	Octadecanoic acid, ethyl ester	2192.5	0.58	GC-MS,Ms
42	Docosane	2197.8	0.74	GC-MS,Ms
43	1-chloro-7-Heptadecene	2206.7	0.18	GC-MS,Ms
44	Benzene, 1,1'-sulfonylbis[4-chloro-	2248	0.07	GC-MS,Ms
45	Methyl dehydroabietate	2361	0.39	GC-MS,Ms
46	Tetracosane	2398.8	1.16	GC-MS,Ms
47	p-Menth-8(10)-en-9-ol	2418	0.05	GC-MS,Ms
48	2,4-bis(1-phenylethyl)phenol	2425.6	1.23	GC-MS,Ms
49	Dehydroabietic acid	2457	1.6	GC-MS,Ms
50	Pentacosane	2497.5	3.18	GC-MS,Ms
51	Bis(2-ethylhexyl) phthalate	2550.6	0.57	GC-MS,Ms
52	Hexacosane	2597	0.33	GC-MS,Ms
53	Heptacosane	2698.6	0.97	GC-MS,Ms
54	Squalene	2831.8	0.74	GC-MS,Ms
55	4-(benzylamino)-1,3-diphenyl-5,6,7,8-tetrahydroquinolin-2(1H)-one	3045	1.98	GC-MS,Ms
56	Vitamin E	3142	0.69	GC-MS,Ms
57	Campesterol	3243.7	0.98	GC-MS,Ms
58	Stigmasterol	3274.7	2.66	GC-MS,Ms
59	γ-Sitosterol	3790	3.67	GC-MS,Ms
Total:		76.27		

a: Retention Indices on RTX-SMS

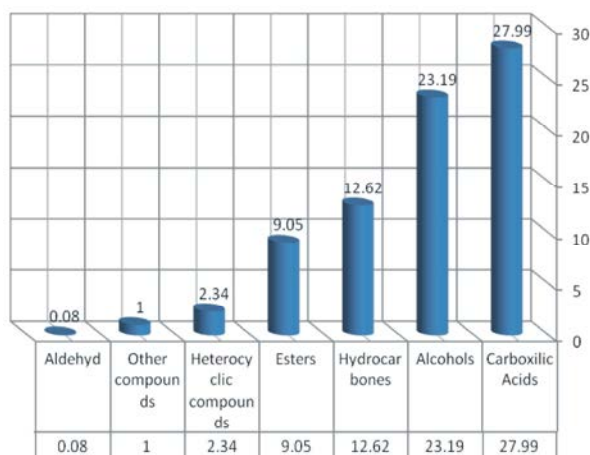


Chart 1: Percentage of compounds find in essential oil of *Spartium junceum*

High resolution gas chromatography-mass spectrometric (HP GC-MS) analysis and Kováts Index values showed that its principal components are the n-Hexadecanoic acid (14.27%), 9,12,15-Octadecatrien-1-ol (13.07%),

Tetradecanoic acid (6.59%) Octadecanoic acid(3.68%) and γ-Sitosterol(3.67%).

chart 1 shows that number of carboxylic acid compounds as the largest group of compound in the n-hexane extract.

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