American-Eurasian Journal of Toxicological Sciences 8 (3): 115-119, 2016 ISSN 2079-2050 © IDOSI Publications, 2016 DOI: 10.5829/idosi.aejts.2016.8.3.1116

Impact of Cobalt on the Growth, Pigmental, Some Biochemical and Enzymatic Characteristics of *Eleusine coracana* **(L.) Gaertn**

^{*I*}M. Duraipandian, ^{*I*}K. Marisamy, ^{*I*}G. Periyanayagi, ^{*I*}R. Sevugaperumal, ²D. Ganesh and ^{*IV*. Ramasubramanian}

¹Department of Botany, Ayya Nadar Janaki Ammal College (Autonomous), Sivakasi 626124, Virudhunagar District, Tamil Nadu, India ²School of Biotechnology, Madurai Kamaraj University, Madurai 625002, Tamil Nadu, India

Abstract: Various pollutants cause different types of pollution in the environment. Among the pollutants, the heavy metals are considered as emerging pollutants in daily life of food crop plants. These heavy metals cause soil and water pollution in our biosphere. Cobalt is one of the heavy metals to cause soil and water pollution in excess. In this study, the seedlings of (finger millet) *Eleusine coracana* (L) Gaertn were treated with various concentrations of cobalt chloride and its impact on the morphometric, pigmental, biochemical and enzymatic characters was studied. After ten days of treatment with different concentrations of cobalt chloride (2, 4, 6, 8, & 10 mM), the growth parameters such as leaf area, fresh weight, dry weight, shoot length, root length were decreased compare to the control. Pigmental characters such as chlorophyll a, chlorophyll b, total chlorophylls and carotenoids content were decreased with increasing the concentrations of cobalt chloride, while the content of anthocyanin was increased of the biochemical parameters the soluble sugar and protein content were found decreased with the increase in the concentrations of cobalt chloride and the contents of free amino acid, proline and leaf nitrate were increased with increasing the concentrations of cobalt chloride. The activities of enzymes such as catalase and peroxidase were increased with increasing the concentration of cobalt chloride and the NRA enzyme activity was decreased. The Atomic Absorption of Spectroscopy (AAS) technique was employed to confirm the presence of cobalt chloride in the treated plants and control. Comparison of the values of treated plants with control reveals that cobalt chloride has seriously affected the finger millet plants.

Key words: Cobalt • Finger millet • Morphological • Antioxidants • Biochemical • AAS

chromium, cadmium, cobalt, lead, mercury and copper etc. enzymes and by substitution of essential metals [3]. Interestingly, small amounts of these elements are actually The activities of several enzymes are also disturbed [4]. necessary for good health, but large amounts of any of In general, the average level of Co in the soil ranges them may cause acute or chronic toxicity. The presence of 30-40 ppm [5] and above that it generates toxicity. heavy metals in the environment is of major concern Plant species vary in their sensitivity to Co. Soil type and because of their toxicity and threat to plant and animal life. soil chemistry greatly influences Co toxicity. One of the Cobalt (Co) is a natural earth element presented in trace most important soil properties is soil acidity and more amount in soil and necessary for normal metabolic acidic the soil, the greater the potential for Co toxicity, at functions in plants, but at higher concentration it is toxic any concentration. During seed germination it causes and may severely interfere with physiological and ultra – structural changes & may cause inhibition in biochemical functions [1]. In excess concentration it growth of plumule & radicals [6] and concentrations caused a marked reduction in growth together with suppressed the seedlings growth and dry weight of the chlorosis and necrosis [2]. Many of these plant responses ragi seedlings [7].

INTRODUCTION to heavy metals are as a result of the inhibition of Our environment has a lot of heavy metals such as metal ions to sulfhydryl groups in the active sites of enzymatic activity caused by the binding of heavy

Corresponding Author: V. Ramasubramanian, Department of Botany, Ayya Nadar Janaki Ammal College (Autonomous), Sivakasi 626124, Virudhunagar District, Tamil Nadu, India.

Both control and experimental plants were allowed to crude enzymes such as peroxidise and catalase. grow in soil containing red, black and sand in the ratio of To assay peroxidise activity, the enzyme extract was 1:1:1. After seven days the seedling of *Eleusine coracana* added to pyrogallol which gets oxidized to a coloured (L) Gaertn were treated with various concentrations of derivative in the presence of hydrogen peroxide (1% (v/v). cobalt chloride such as 2, 4, 6, 8 & 10mM. After ten days The amount of purpurogallin formed during the reaction of cobalt treatment various morphometric, pigmental, was assayed spectroscopically [13]. To 2 mL of pyrogallol

and root length are measured by scale and expressed in extract was added. Then, absorbance was set to zero at the leaf area was measured by conventional graphical $\,$ 420nm. To this, 0.5 mL of H₂O₂ (1% (v/v)) was added. method by drawing the outline of the leaf in a graph sheet Then, the content was thoroughly mixed and the and counting the small and big squares with in it and absorbance was measured using systronics model no. 106 represented in cm². The fresh weight and dry weight of spectrophotometer. The difference in the absorbance at

chlorophyll b, total chlorophyll and carotenoids are measured by conventional method [8]. To assay the catalase activity, 3 mL of phosphate

sugars are measured by centrifugation after treating the extract [14]. The reaction mixture was incubated at 25° C for tissue use tri-chloro acetic acid and anthrone as the 1 minute. The reaction was terminated by the addition of indicator the total soluble sugars. The amino acid content was measured by homogenation of leaf tissue in ethanol 0.01 N KMNO₄. The end point was the persistence of pink and ninhydrine is the indicator of amino acid [9]. The colour at least for 15 seconds. The catalase activity was protein content was measured by centrifugation leaf tissue in tri-chloro acetic acid, folin phenol with alkaline mg protein. copper mixture [10]. The proline was estimated by The heavy metal accumulation of the experimental centrifugation of leaf tissue in sulphosalicylic acid, plants was analyzed at the end of the plant life. Cobalt glacial acetic acid and the proline was separated in the concentrations in plants were analyzed using the method toluene [11]. [15].

assayed according to Jaworski *et al* 1971[12] method with oven at 160ºC for 40 minutes and digested in a mixture of modification. Fresh leaf material (100mg) was incubated in nitric acid and perchloric acid (10:1). Then the solution scintillation vials containing 5 mL of incubation medium was centrifuged at 5000 rpm for 5 minutes and double composed of, 100mM KH₂PO₄.KOH buffer pH 7.5, 200mM filtered with Whatmann filter paper no.4 and the filtrate KNO₃, 1%(v/v) n-Propanol and 0.1%(v/v) Triton X 100. was analyzed for cobalt concentration by Atomic Incubation was carried out in dark for one hour at room Absorption Spectrometry (Shimadzu Model AA – 6300), temperature with occasional shaking. Aliquots of 0.5 mL available in the Science Instrumentation Centre of Ayya was taken from the vials and analysed for nitrite after 1 Nadar Janaki Ammal College (Autonomous), Sivakasi, hour incubation. To 0.5 mL distilled water, 1 mL of 3% Tamilnadu. sulphanilamide and 1 mL of N-1-N (Naphthyl ethylenediamine dihydrochloride) were added in quick succession. **Statistical Analysis:** For the morphometric characters the

phosphate buffer (pH 6.0) and filtered through a three package, Origin – version 7.0.

MATERIALS AND METHODS layered cheese cloth and spun at 3000 rpm for 30 minutes. The supernatant obtained was served as the source for

biochemical and enzymatic characters were analyzed. phosphate buffer (0.058 M pyrogallol dissolved in 0.1 M The morphometric characters such as shoot length phosphate buffer pH 6.0), an aliquot of 0.1 mL of enzyme the seedlings was obtained using an electronic balance. an interval of 20 seconds for a period of 3 minutes was The pigmental characters such as chlorophyll a, measured. The peroxidise activity was expressed as moles of H₂O₂ reduced per unit enzyme per unit time.

The biochemical characters such as total soluble buffer was added to 1 mL of H_2O_2 and 1 ml of enzyme 1 mL of H_2SO_4 . The reaction mixture was titrated against expressed in micromoles H_2O_2 catalysed per unit time per

The *In vivo* Nitrate Reductase (NR) activity was The plant sample as a whole was washed, dried in

Fifteen minutes were allowed for colour formation and average value of ten independent samples and for the absorbance was measured at 540 nm. biochemical readings and enzymatic characters the The nitrite was estimated with the help of a standard average value of five were considered. The data were nitrite curve. The preparation of peroxidase and catalase reported as mean \pm SE and the percent activity was enzyme extracts is, the leaves of experimental plants represented in the parentheses. Statistical analysis (One weighing about 1 g was ground in 5 mL of 100 Mm way ANOVA – Turkey test) was done using the statistical

such as root length, shoot length, leaf area, fresh weight enzymes such as peroxidase and catalase increased and dry weight were decreased with increasing the with the increase in the metal concentration. But concentration of cobalt chloride (Table 1). The levels of nitrate reductase activity decreased (Table 4). The metal chlorophyll and carotenoids decreased but the level of content of the treated and control seedlings of *Eleusine* anthocyanin increased (Table 2). Similarly the total *coracana* (L) Gaertn. was finally estimated by AAS study soluble sugars and protein also showed a declining trend Table (5).

RESULTS with increasing the concentration of cobalt chloride The results showed that the morphometric characters and proline (Table 3) and the activities antioxidant (Table 3). In contrary, the leaf nitrate, free amino acid

Table 1: Impact of various concentration of cobalt chloride on the morphometric characteristics of *Eleusine coracana* (L.) Gaertn.

Cobalt concentration (mM)	Shoot Length (cm)	Root Length(cm)	Leaf area $(cm2)$	Fresh weight (g)	Dry weight (g)
Control	$17.16 \pm 0.545(100)$	$9.46\pm0.202(100)$	$4.33\pm0.145(100)$	$0.25 \pm 0.030(100)$	$0.12\pm0.01(100)$
2	$15.26 \pm 0.497(88.93)$	7.933 \pm 0.202 a* (83.81)	4.033 ± 0.346 a* (93.08)	0.21 ± 0.014 a* (83.14)	0.10 ± 0.014 a* (84.21)
$\overline{4}$	$13.46 \pm 0.317(78.45)$	6.933 ± 0.145 a* (73.24)	3.533 ± 0.088 a [*] (81.54)	0.14 ± 0.012 a [*] (55.86)	0.08 ± 0.012 a* (65.79)
6	$12.26 \pm 0.272(71.46)$	6.433 ± 0.088 a* (67.96)	9.85 ± 0.263 a*(76.78)	0.10 ± 0.008 a [*] (41.57)	0.07 ± 0.008 a [*] (60.52)
8	$11.66 \pm 0.145(67.97)$	5.733 ± 0.666 a* (60.57)	9.52 ± 0.216 a [*] (74.21)	0.09 ± 0.012 a* (37.67)	0.05 ± 0.008 a [*] (44.74)
10	$10.10\pm0.103(58.85)$	4.766 ± 0.033 a* (50.36)	8.02 ± 0.174 a [*] (62.52)	0.063 ± 0.008 a* (24.68)	0.04 ± 0.005 a [*] (31.58)

Eleusine coracana (L.) Gaertn.

Values are an average of ten observations. Values in parenthesis are percentage activity with respect to control. Mean±SE a – 2mM to 10mM Concentrations compared with control, *Significant ($P = 0.05 -$ Turkey test).

Values are an average of ten observations. Values in parenthesis are percentage activity with respect to control. Mean \pm SE a – 2mM to 10mM Concentrations compared with control, *Significant ($P = 0.05 -$ Turkey test).

Table 3: Impact of various concentration of cobalt chloride on the biochemical characteristics of *Eleusine coracana* (L.) Gaertn.

Cobalt concentration (mM)	Total soluble sugar	Total soluble protein	Amino acid	Proline	Leaf nitrate
	$(mg/g$ LFW)	(mg/g LFW)	$(\mu \text{ Mole/g LFW})$	$(mg/g$ LFW)	(mg/g LFW)
Control	$5.52 \pm 0.178(100)$	$1.22 \pm 0.037(100)$	$16.36 \pm 0.306(100)$	$2.91 \pm 0.036(100)$	$0.11 \pm 0.001(100)$
2	4.923 ± 0.031 a*(89.08)	1.06 ± 0.027 a* (87.32)	18.41 ± 0.573 a* (112.6)	3.62 ± 0.074 a* (124.2)	0.15 ± 0.004 a [*] (140)
$\overline{4}$	3.78 ± 0.051 a [*] (68.40)	0.91 ± 0.035 a [*] (74.64)	20.65 ± 0.430 a [*] (126.2)	4.30 ± 0.050 a [*] (147.6)	0.17 ± 0.002 a [*] (160.6)
6	2.91 ± 0.028 a [*] (52.65)	0.79 ± 0.013 a* (65.24)	26.25 ± 0.473 a*(160.4)	5.14 ± 0.042 a [*] (176.4)	0.22 ± 0.002 a*(204.8)
8	2.25 ± 0.053 a*(40.83)	0.71 ± 0.018 a [*] (58.69)	36.61 ± 0.786 a*(210.4)	5.61 ± 0.034 a [*] (192.5)	0.27 ± 0.001 a*(247.7)
10	1.53 ± 0.103 a* (27.80)	0.64 ± 0.013 a [*] (52.56)	49.61 \pm 0.620 a*(297.6)	5.78 ± 0.042 a*(200.4)	0.34 ± 0.003 a*(306)

Values are an average of ten observations. Values in parenthesis are percentage activity with respect to control. Mean±SE a – 2mM to 10mM Concentrations compared with control, *Significant ($P = 0.05 -$ Turkey test).

Table 4: Impact of various concentration of cobalt chloride on the enzymes activity of *Eleusine coracana* (L.) Gaertn.

Cobalt concentration (mM)	NR Activity (μ Mole/g LFW)	Peroxidase (µ Mole/g LFW)	Catalase (μ Mole/g LFW)
Control	$10.77\pm0.069(100)$	$6.88 \pm 0.030(100)$	$1.55\pm0.058(100)$
2	7.24 ± 0.145 a* (67.53)	7.00 ± 0.016 a [*] (101.8)	2.15 ± 0.058 a* (138.6)
$\overline{4}$	6.52 ± 0.154 a* (60.88)	7.19 ± 0.029 a* (104.5)	2.86 ± 0.117 a [*] (187.1)
6	5.91 ± 0.097 a* (54.15)	7.99 ± 0.020 a [*] (116.1)	3.66 ± 0.117 a* (224.3)
8	5.38 ± 0.048 a* (50.26)	8.68 ± 0.039 a* (126.2)	3.73 ± 0.058 a* (247.1)
10	4.69 ± 0.042 a* (44.56)	10.01 ± 0.067 a* (144.9)	4.22 ± 0.080 a [*] (271.4)

Table 5: AAS study indicating the content of cobalt in *Eleusine coracana* (L.) Gaertn.

SAMPLE	AAS in ppm
SAMPLE 1 (Control)	0.8818 ppm
SAMPLE 2 (2 mM)	3.2406ppm
SAMPLE 3 (4 mM)	4.1155ppm
SAMPLE 4 (6 mM)	4.7864ppm
SAMPLE 5 (8 mM)	5.5056ppm
SAMPLE 6 (10 mM)	6.7484ppm

Values are an average of ten observations. Values in parenthesis are

percentage activity with respect to control. Mean±SE a – 2mM to 10mM Concentrations compared with control, *Significant ($P = 0.05$ - Turkey test).

DISCUSSION

Heavy metal is known to enhance the level of the enzyme chlorophyllase that brings about the degradation of the chlorophyll [16]. The reduction in sugar contents may be attributed to reduction in chlorophyll contents of the leaf and also a decline in protein. This change might have already affected the photosynthetic activity of the plant and hence the reduction in contents [17].

Accumulation of proline has been frequently used as biochemical marker for water stress in plants [18]. In stress condition the inhibition of growth of cells, leaves and the whole plant were accompanied by an accumulation of nitrate in plant tissue particularly in leaves [19]. The leaf nitrate content was found to be more in cobalt treated plants paralleling with the reduction in nitrate reductase activity. The peroxidase activity was found to be increased with increasing the concentration of the cobalt chloride and cause a major impact on the chlorophyll degradation. Heavy metals are well known, to generate a large quantity of reactive oxygen species (ROS) in plants that may oxidize protein, lipids and nucleic acids resulting in the abnormalities at the level of the cell [20].

To maintain metabolic functions under stress conditions, the balance between generation and degradation of ROS is required; for this purpose plants have well equipped antioxidant system that include antioxidant enzymes (superoxide dismutase, catalase, peroxidase and glutathione reductase) and non- enzymatic low molecular weight antioxidants (glutathione, proline, carotenoids, tocopherols etc.) [21]. An increase in the amino acid and proline content after match, sugar industrial effluent treatment has already been reported [22][23]. Catalase is an antioxidant and scavenging enzyme, found to be increased with the increasing concentration of cobalt chloride. Both the catalase and peroxidase catalyse the degradation of H2O2, which is a natural metabolite and also toxic to plants [24].

CONCLUSION

The results of present investigation inferred from the comparison of the values given in parenthesis of treated plants with control reveals that cobalt chloride has seriously affected the finger millet plants.

ACKNOWLEDGEMENT

The authors are thankful to the principal and management of Ayya Nadar Janaki Ammal College (Autonomous), Sivakasi for the facilities provided.

REFERENCES

- 1. Jayakumar, K. and P. Vijayarengan, 2006. Alterations in the carbohydrate metabolism of *Vigna mungo* (L.) Hepper as affected by cobalt stress. Ind. J. Environ. Ecoplan., 12: 693-696.
- 2. Vanselow, A.P., 1966. Cobalt in diadnostic criteria of plants and soils, ed. HD Chapman, Riverside, CA: University of California Division of Agricultural Science, pp: 142-156.
- 3. Van Assche, F. and H. Clijster, 1990. Effects of metals on enzyme activity in plant. Plant Cell Environ., 13: 195-206.
- 4. Shalygo, N.V., N.V. Shalygo, N.V. Kolesnikova, V.V.V. Voronetskaya and N.G. Averina, 1999. Effect of Mn2+, Fe2+, Co2+ and Ni2+ on chlorophyll accumulation and early stages of chlorophyll formation in greening barley seedlings. Russ. J. Plant Physiol., 46: 496-501.
- 5. Kabata-Pendias, A. and H. Pendias, 1991. Trace elements in soils and plants. CRC, Boca Raton FL, pp: 276-285.
- 6. Ayaz, F. and A. Kadioglu, Effects of heavy metals (Zn, Cd, Cu, Hg) on the soluble protein bands of germinating *Lens esculenta* L. seeds. Turkish J. Bot., 21: 85-88.
- 7. Jeyakumar, K., C.A. Jaleel and M.M. Azooz, 2008. Impact of cobalt on germination and seedling growth of *Eleusine coracana* L. and *Oryza sativa* L. under hydroponic culture. Global J. Molecular Sciences, 3(1): 18-20.
- 8. Wellburn, A.R. and H. Lichtenthaler, 1984. In Advances in photosynthesis Research, (ed.Sybesma) Martinus Nijhoff, Co. The Hague., II: 9-12.
- 9. Jayaraman, J., 1981. Laboratory manual in Biochemistry, Willey-Eastern Company Limited, Madras, pp: 1-65.
- phenol reagent. J. Bio. Chem., 193: 262-275. Physiol., 138: 534-538.
- 11. Bates, L.S., R.P. Waldren and I.D. Teare, 1973. Rapid 19. Sinha and Nicholas D.J.D., 1981. In the physiology
- 12. Jaworski, E.G., 1971. Nitrate reductase assay in intact London, pp: 145- 169. plant tissues. Biochem. Biophy. Res. Commun., 20. Sanita di T.L. and R. Gabbrielli, 1999. Response
- 13. Addy, S.K. and R.N. Goodman, 1972. Polyphenol 41: 105-130. oxidase and peroxidise activity in apple leaves 21. Schutzendubell, A. and A. Polle, 2002. Plant response
- 14. Kar, M. and D. Mishra, 1976. Catalase, peroxidase 53: 1351-1365. and polyphenol oxidase activities during rice leaf 22. Jeyarathi, K.P. and V. Ramasubramanian, 2002.
- Botany Research International, 2(4): 310-314. Agrobios Publication, Jodhpur, pp: 245-253.
- 20: 50-55. 24(17 & 18): 2241-2249.
- germination and seedling development of groundnut (*Arachis hypogea*,L.). Envirion. Biol., 2: 187-189.
- 10. Lowry, O.H., N.J. Rosenbury, A.L. Farr and 18. Alia, P. and P.P. Saradhi, 1991. Proline R.J. Randall, 1951. Protein measurement with folin accumulation under heavy metal stress. J. Plant
	- determination of the proline in water stress studies. and biochemistry of drought resistance in plants Plant and Soil, 39: 205-208. (ed. Paleg,L.G. and Aspinall,D.) Academic Press,
	- 43: 1274-1279. of cadmium in higher plant. Environ. Exp. Bot.,
	- inoculated with a virulent or avirulent strain for to abiotic stresses: heavy metal-induced oxidative *Erwinia amylovora*. Indian Phytopath., 25: 575-579. stress and protection by mycorrhization. J. Exp. Bot.,
- senescence. Plant Physiol., 57: 315-319. Analysis of sugar mill Effluents and its impact on the 15. Jayakumar, K. and Cheruth Abdul Jaleel, 2009. growth and biochemical characteristics of Uptake and Accumulation of Cobalt in Plants: a *Abelmascus esculentus* (L) Mediakus. *In: Recent* Study Based on Exogenous Cobalt in Soybean. *Trends in Biotechnology* (Ed.Harikumar,V.S)
- 16. Reddy, M.P. and A.B. Vora, 1986. Changes in 23. Ramasubramanian, V., V. Ravichandran and pigment composition, hill reaction and saccharide N. Kannan, 1993. Analysis of industrial effluents and metabolism in bajra (*Pennisetum typhoides* S&H) their impact on the growth and metabolism of leaves under NaCl salinity. Photosynthetica, *Phaseolus mungo*, L. Commun. Soil. Sci. Plant. Anal.,
- 17. Swaminathan, K., L.J. Arjunan and R. Gurusamy, 24. Balasinha, D., 1982. Regulation of peroxidase in 1998. Effect of glucose factory effluents on the seed higher plants. Ann. Rev. Plant. Physiol., 25: 225-228.